### organic papers

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#### Key indicators

Single-crystal X-ray study T = 273 K Mean  $\sigma$ (C–C) = 0.003 Å R factor = 0.034 wR factor = 0.100 Data-to-parameter ratio = 7.4

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# The title compound, $C_{11}H_{17}NO_2$ , was obtained by hydrogenolysis of (3S,4S)-4-(dibenzylamino)-5-phenylpentane-1,3-diol. In the crystal structure, intermolecular $O-H\cdots O$ , $O-H\cdots N$ and $N-H\cdots O$ hydrogen bonds link the molecules into a sheet parallel to the *ab* plane.

(3S,4S)-4-Amino-5-phenylpentane-1,3-diol

#### Comment

The title compound, (I), is a key intermediate used to synthesize recycling chiral ligands for asymmetric catalysts such as pyridinyloxazolines. X-ray analysis of (I) reveals that the amino group is in a pyramidal configuration with the sum of the bond angles around N1 being  $324^{\circ}$ . Symmetry-related molecules are linked *via* O-H···O, O-H···N and N-H···O intermolecular hydrogen bonds into a sheet parallel to the *ab* plane.



#### **Experimental**

To a solution of (3S,4S)-4-(dibenzylamino)-5-phenylpentane-1,3-diol (1125 mg, 3 mmol) in methanol (20 ml), ammonium formate (1134 mg, 18 mmol) and palladium on carbon (10% Pd) (253 mg) were added and the mixture heated to 353 K. It was stirred for 3 h at 353 K (Stuk *et al.*, 1994). After completion of the reaction, the catalyst was filtered off, the filtrate was concentrated under reduced pressure to give the crude product. It was purified by column chromatography over silica gel (ethyl acetate-methanol, 20:1  $\nu/\nu$ ) to yield compound (I) (527 mg). Single crystals were obtained by recrystallization from a mixture of ethyl acetate and hexane.

Crystal data	
C <sub>11</sub> H <sub>17</sub> NO <sub>2</sub>	$D_x = 1.236 \text{ Mg m}^{-3}$
$M_r = 195.26$	Mo $K\alpha$ radiation
Monoclinic, P2 <sub>1</sub>	Cell parameters from 2218
a = 7.960 (3)  Å	reflections
b = 5.4531 (17)Å	$\theta = 2.6-28.5^{\circ}$
c = 12.142 (4) Å	$\mu = 0.09 \text{ mm}^{-1}$
$\beta = 95.379 \ (6)^{\circ}$	T = 273 (2) K
V = 524.7 (3) Å <sup>3</sup>	Block, colourless
<i>Z</i> = 2	$0.56 \times 0.28 \times 0.18 \ \mathrm{mm}$
Data collection	
Bruker APEX area-detector	1019 independent reflections
diffractometer	1006 reflections with $I > 2\sigma(I)$
$\varphi$ and $\omega$ scans	$R_{\rm int} = 0.021$
Absorption correction: multi-scan	$\theta_{\rm max} = 25.0^{\circ}$
(SADABS; Bruker, 2001)	$h = -9 \rightarrow 8$
$T_{\min} = 0.959, \ T_{\max} = 0.987$	$k = -6 \rightarrow 6$
2647 measured reflections	$l = -14 \rightarrow 13$

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#### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0762P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.034$	+ 0.0605P]
$wR(F^2) = 0.100$	where $P = (F_0^2 + 2F_c^2)/3$
S = 0.98	$(\Delta/\sigma)_{\rm max} = 0.001$
1019 reflections	$\Delta \rho_{\rm max} = 0.11 \ {\rm e} \ {\rm \AA}^{-3}$
137 parameters	$\Delta \rho_{\rm min} = -0.15 \text{ e } \text{\AA}^{-3}$
All H-atom parameters refined	

#### Table 1

Selected geometric parameters (Å, °).

N1-C4	1.465 (3)	C5-C6	1.500 (3)
O1-C1	1.406 (3)	C6-C7	1.370 (3)
O2-C3	1.416 (3)	C6-C11	1.375 (3)
C1-C2	1.496 (3)	C7-C8	1.374 (4)
C2-C3	1.508 (3)	C8-C9	1.369 (4)
C3-C4	1.523 (3)	C9-C10 1.361	
C4-C5	1.523 (3)	C10-C11	1.372 (3)
01 C1 C2	100.72 (17)	C7 C6 C11	1185(2)
C1 - C1 - C2	109.75(17) 112.16(10)	$C_{7} = C_{0} = C_{11}$	110.3(2) 121.6(2)
C1 - C2 - C3	112.10(19) 111.51(19)	$C_{1} = C_{0} = C_{3}$	121.0(2) 110.0(2)
02 - C3 - C2	111.31(10) 111.92(10)	$C_{11} = C_{0} = C_{0}$	119.9(2)
02 - 03 - 04	111.82 (18)	$c_{0} = c_{1} = c_{8}$	120.8 (2)
$C_2 = C_3 = C_4$	112.62 (19)	09-08-07	119.9 (2)
N1 - C4 - C3	113.26 (16)	C10 - C9 - C8	119.9 (2)
N1-C4-C5	108.44 (16)	C9-C10-C11	120.0 (2)
C3-C4-C5	111.89 (19)	C10-C11-C6	120.8 (2)
C6-C5-C4	114.26 (17)		
C1-C2-C3-C4	164.94 (18)	C3-C4-C5-C6	62.5 (3)
C2-C3-C4-N1	59.3 (3)	C4-C5-C6-C7	-90.5(3)
O2-C3-C4-C5	55.7 (2)	C4-C5-C6-C11	89.1 (2)
N1-C4-C5-C6	-171.8 (2)		( )

## Table 2Hydrogen-bond geometry (Å, °).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D{\cdots}A$	$D - H \cdots A$
O1-H1···N1 <sup>i</sup>	0.82	1.96	2.763 (2)	167
O2-H2··· $O1$ <sup>ii</sup>	0.82	1.99	2.805 (2)	172
$N1-H1N\cdotsO1^{ii}$	0.87(1)	2.43 (1)	3.288 (3)	168 (3)
$N1 - H2N \cdots O1^{iii}$	0.87 (1)	2.60 (1)	3.456 (3)	167 (3)

Symmetry codes: (i) x + 1, y, z; (ii)  $-x + 2, y - \frac{1}{2}, -z$ ; (iii)  $-x + 2, y + \frac{1}{2}, -z$ .

Amino H atoms were located in a difference map and were refined with an N-H distance restraint of 0.87 (1) Å. All other H atoms were positioned geometrically (C-H = 0.93, 0.97 or 0.98 Å for phenyl, methylene or tertiary H atoms, respectively, and O-H = 0.82 Å) and were included in the refinement in the riding-model approximation, with  $U_{\rm iso}(\rm H) = 1.2U_{eq}$  (carrier atom). In the absence of significant anomalous scattering effects, Friedel pairs were merged; the absolute configuration was assumed from the synthesis.



Figure 1
ORTEPIII (Farrugia, 1997) plot of (I). Displacement ellipsoids are drawn
at the 50% probability level. H atoms are drawn as spheres of arbitrary
radii.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINT* (Bruker, 2001); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPIII* (Farrugia, 1997); software used to prepare material for publication: *SHELXL97*.

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